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## Structure Reports

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**(2E,4E)-Ethyl 5-(phenylsulfonyl)penta-2,4-dienoate**Ulaganathan Sankar,<sup>a</sup> V. Sabari,<sup>b</sup> G. Suresh,<sup>b</sup>  
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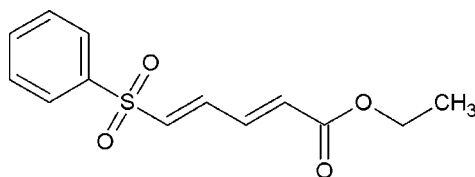
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Key indicators: single-crystal X-ray study;  $T = 298$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  
 $R$  factor = 0.043;  $wR$  factor = 0.147; data-to-parameter ratio = 18.8.

In the title compound,  $\text{C}_{13}\text{H}_{14}\text{O}_4\text{S}$ , both  $\text{C}=\text{C}$  double bonds adopt an *E* conformation. In the crystal, molecules are linked into centrosymmetric  $R_2^2(14)$  dimers *via* pairs of  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds.

## Related literature

For the biological activity of phenyl sulfonyl-containing compounds see: De-Benedetti *et al.* (1985). For a related structure, see: Li (2011).



## Experimental

## Crystal data

$\text{C}_{13}\text{H}_{14}\text{O}_4\text{S}$   
 $M_r = 266.30$   
Triclinic,  $P\bar{1}$

$a = 6.2525$  (3) Å  
 $b = 7.8889$  (4) Å  
 $c = 14.5049$  (7) Å

$\alpha = 82.828$  (3)°  
 $\beta = 87.261$  (2)°  
 $\gamma = 72.426$  (2)°  
 $V = 676.69$  (6) Å<sup>3</sup>  
 $Z = 2$

Mo  $K\alpha$  radiation  
 $\mu = 0.24$  mm<sup>-1</sup>  
 $T = 298$  K  
 $0.45 \times 0.38 \times 0.15$  mm

## Data collection

Bruker APEXII KappaCCD diffractometer  
Absorption correction: multi-scan (SADABS; Bruker, 2008)  
 $T_{\min} = 0.899$ ,  $T_{\max} = 0.965$

8948 measured reflections  
3084 independent reflections  
2621 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.043$   
 $wR(F^2) = 0.147$   
 $S = 1.87$   
3084 reflections

164 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.29$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.30$  e Å<sup>-3</sup>

**Table 1**  
Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$                          | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--|-------|-------------|-------------|---------------|
| $\text{C7}-\text{H7}\cdots\text{O3}^i$ | 0.93  | 2.32        | 3.212 (2)   | 161           |

Symmetry code: (i)  $-x + 2, -y, -z + 1$ .

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2009).

SA thanks the UGC, India, for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5831).

## References

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## supplementary materials

*Acta Cryst.* (2012). E68, o1093 [doi:10.1107/S1600536812009907]

**(2E,4E)-Ethyl 5-(phenylsulfonyl)penta-2,4-dienoate**

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**Comment**

Phenyl sulfonyl containing compounds show a wide range of biological properties (De-Benedetti *et al.*, 1985).

Fig. 1. shows a displacement ellipsoid plot of the title compound. Both C=C double bonds display an E configuration. The title molecule exhibits structural similarities with the already reported related structure (Li, 2011). The dihedral angle between two planes (C5—C6—S1—O1) and (C1—C6—S1—O2) is 37.32 (6)°. The crystal packing is stabilized by C—H···O intermolecular interactions. The molecules are linked into centrosymmetric  $R_2^2(14)$  dimers via C7—H7···O3 hydrogen bonds (Table 1). The packing of the compound is shown in (Fig. 2).

**Experimental**

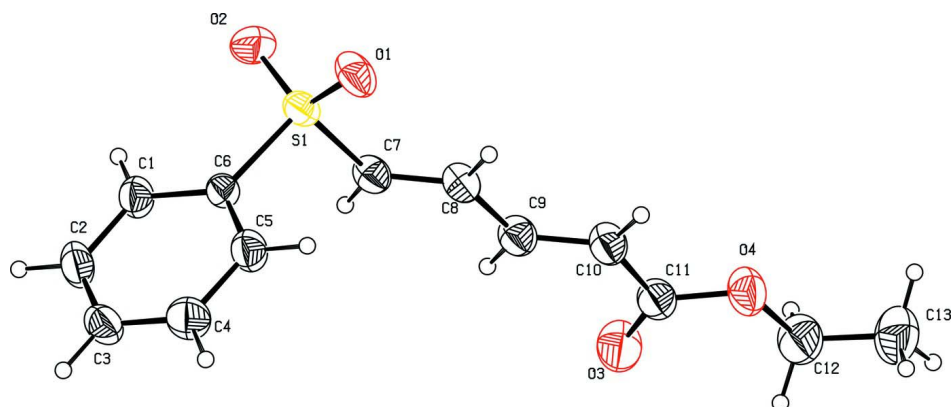
LIHMDS (8 ml, 8.4 mmol, 1.06 molar solution in THF) was added drop wise to a 0 °C cooled solution of bisphenyl sulfonyl methane (1 g, 3.4 mmol) in dried THF (15 ml) under argon atmosphere. The reaction mixture was stirred at same temp for 1 h, and then trans ethyl 4-bromo coronate (0.71 g, 3.7 mmol) in dry THF (5 ml) was added dropwise and the reaction mixture was allowed to come to RT and it was stirred under argon atmosphere for 16 h. The reaction mixture was quenched by adding saturated NH<sub>4</sub>Cl (20 ml) and then extracted with ethyl acetate (2x20 ml) and washed with water (2x20 ml) and sat brine (20 ml). Then, the organic layer was dried over MgSO<sub>4</sub>. Evaporation of the solvent under vacuum furnished the crude product, the residue was chromatographed (25% ethyl acetate in hexanes) to give analytically pure (2E, 4E)-ethyl 5-(phenyl sulfonyl)penta-2,4-dienote (0.673 g. yield 75%) as a colorless solid.

**Refinement**

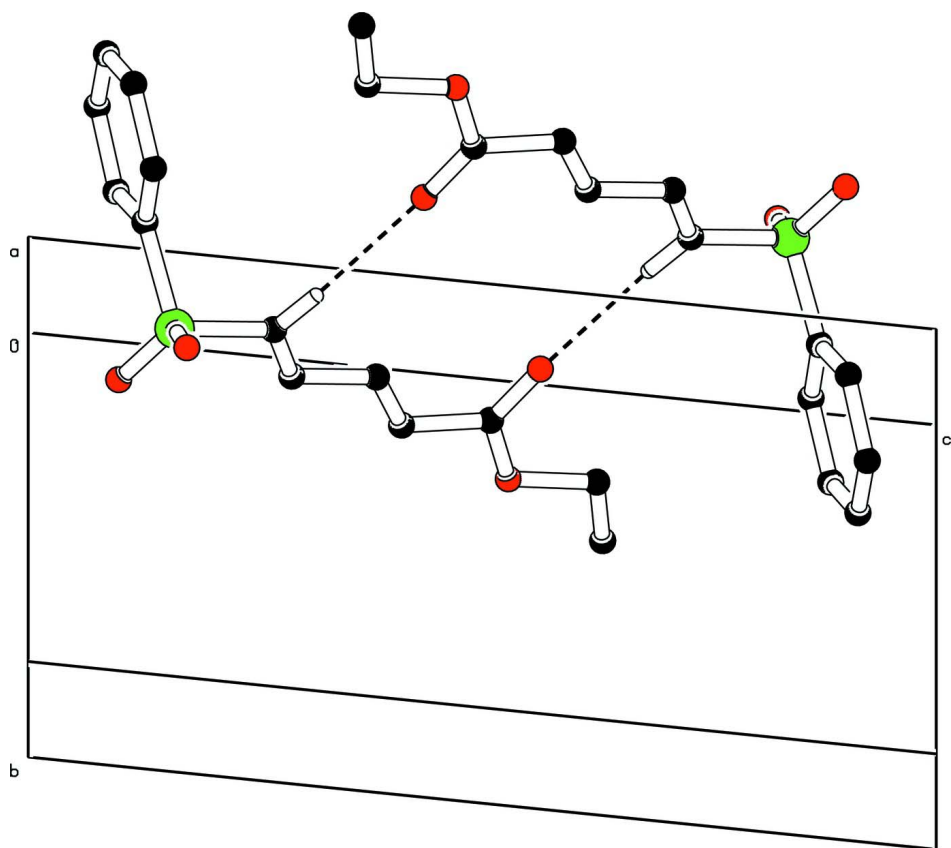
Hydrogen atoms were placed in calculated positions with C—H ranging from 0.93 Å to 0.97 Å and refined using a the riding model with fixed isotropic displacement parameters:  $U_{iso}(H) = 1.5 U_{eq}(C)$  for the methyl group and  $U_{iso}(H) = 1.2 U_{eq}(C)$  for other groups.

**Computing details**

Data collection: *APEX2* (Bruker, 2008); cell refinement: *SAINTE* (Bruker, 2008); data reduction: *SAINTE* (Bruker, 2008); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008) and *PLATON* (Spek, 2009).

**Figure 1**

The molecular structure of the title compound with the atom numbering scheme. Displacement ellipsoids are shown at 30% probability level. The H atoms are presented as a small spheres of arbitrary radius. Related atoms have symmetry code: (i)  $-x + 2, -y, -z + 1$ .

**Figure 2**

A view of the crystal packing. H atoms not involved in hydrogen bonding(dashed lines)have been omitted for clarity.

(2*E*,4*E*)-Ethyl 5-(phenylsulfonyl)penta-2,4-dienoate

Crystal data

|                                |   |
|--------------------------------|---|
| $C_{13}H_{14}O_4S$             | $Z = 2$   |
| $M_r = 266.30$                 | $F(000) = 280$  |
| Triclinic, $P\bar{1}$          | $D_x = 1.307 \text{ Mg m}^{-3}$                         |
| $a = 6.2525 (3) \text{ \AA}$   | Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$ |
| $b = 7.8889 (4) \text{ \AA}$   | Cell parameters from 5946 reflections                   |
| $c = 14.5049 (7) \text{ \AA}$  | $\theta = 2.7\text{--}28.3^\circ$                       |
| $\alpha = 82.828 (3)^\circ$    | $\mu = 0.24 \text{ mm}^{-1}$                            |
| $\beta = 87.261 (2)^\circ$     | $T = 298 \text{ K}$                                     |
| $\gamma = 72.426 (2)^\circ$    | Triclinic, colourless                                   |
| $V = 676.69 (6) \text{ \AA}^3$ | $0.45 \times 0.38 \times 0.15 \text{ mm}$               |

Data collection

|  |  |
|--|--|
| Bruker APEXII KappaCCD diffractometer                    | 8948 measured reflections  |
| Radiation source: fine-focus sealed tube                 | 3084 independent reflections   |
| Graphite monochromator                                   | 2621 reflections with $I > 2\sigma(I)$                                 |
| Detector resolution: $15.9948 \text{ pixels mm}^{-1}$    | $R_{\text{int}} = 0.020$   |
| $\omega$ and $\varphi$ scans                             | $\theta_{\text{max}} = 28.4^\circ$ , $\theta_{\text{min}} = 2.7^\circ$ |
| Absorption correction: multi-scan (SADABS; Bruker, 2008) | $h = -8 \rightarrow 8$   |
| $T_{\text{min}} = 0.899$ , $T_{\text{max}} = 0.965$      | $k = -10 \rightarrow 9$  |
|  | $l = -19 \rightarrow 19$   |

Refinement

|  |  |
|--|--|
| Refinement on $F^2$  | Hydrogen site location: inferred from neighbouring sites |
| Least-squares matrix: full                                     | H-atom parameters constrained                            |
| $R[F^2 > 2\sigma(F^2)] = 0.043$                                | $w = 1/[\sigma^2(F_o^2) + (0.0551P)^2]$                  |
| $wR(F^2) = 0.147$  | where $P = (F_o^2 + 2F_c^2)/3$                           |
| $S = 1.87$   | $(\Delta/\sigma)_{\text{max}} = 0.001$                   |
| 3084 reflections   | $\Delta\rho_{\text{max}} = 0.29 \text{ e \AA}^{-3}$      |
| 164 parameters   | $\Delta\rho_{\text{min}} = -0.30 \text{ e \AA}^{-3}$     |
| 0 restraints   | Extinction correction: SHELXL97 (Sheldrick, 2008)        |
| Primary atom site location: structure-invariant direct methods | Extinction coefficient: 0.0173 (18)                      |
| Secondary atom site location: difference Fourier map           |  |

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

|    | $x$        | $y$         | $z$          | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|------------|-------------|--------------|----------------------------------|
| C1 | 1.5671 (3) | -0.0639 (2) | 0.13787 (12) | 0.0577 (4)                       |
| H1 | 1.6549     | -0.0001     | 0.1586       | 0.069*                           |

|      |             |              |              |              |
|------|-------------|--------------|--------------|--------------|
| C2   | 1.6638 (3)  | -0.2359 (3)  | 0.11608 (12) | 0.0670 (5)   |
| H2   | 1.8179      | -0.2881      | 0.1219       | 0.080*       |
| C3   | 1.5361 (3)  | -0.3300 (2)  | 0.08618 (12) | 0.0634 (5)   |
| H3   | 1.6035      | -0.4465      | 0.0726       | 0.076*       |
| C4   | 1.3103 (3)  | -0.2555 (3)  | 0.07586 (14) | 0.0714 (5)   |
| H4   | 1.2246      | -0.3206      | 0.0549       | 0.086*       |
| C5   | 1.2088 (3)  | -0.0820 (2)  | 0.09680 (12) | 0.0612 (4)   |
| H5   | 1.0550      | -0.0299      | 0.0895       | 0.073*       |
| C6   | 1.3387 (2)  | 0.0124 (2)   | 0.12854 (9)  | 0.0446 (3)   |
| C7   | 1.0978 (3)  | 0.1812 (2)   | 0.27041 (10) | 0.0534 (4)   |
| H7   | 1.1952      | 0.1132       | 0.3169       | 0.064*       |
| C8   | 0.8815 (3)  | 0.2378 (2)   | 0.29034 (11) | 0.0545 (4)   |
| H8   | 0.7812      | 0.3023       | 0.2441       | 0.065*       |
| C9   | 0.7980 (3)  | 0.2008 (2)   | 0.38347 (11) | 0.0585 (4)   |
| H9   | 0.9013      | 0.1296       | 0.4271       | 0.070*       |
| C10  | 0.5861 (3)  | 0.2604 (2)   | 0.41125 (11) | 0.0603 (4)   |
| H10  | 0.4768      | 0.3307       | 0.3697       | 0.072*       |
| C11  | 0.5253 (3)  | 0.2140 (3)   | 0.50916 (12) | 0.0616 (4)   |
| C12  | 0.2457 (4)  | 0.2713 (3)   | 0.62613 (15) | 0.0902 (7)   |
| H12A | 0.2830      | 0.1440       | 0.6466       | 0.108*       |
| H12B | 0.3229      | 0.3242       | 0.6655       | 0.108*       |
| C13  | 0.0051 (4)  | 0.3538 (4)   | 0.63268 (17) | 0.1022 (8)   |
| H13A | -0.0313     | 0.4779       | 0.6084       | 0.153*       |
| H13B | -0.0420     | 0.3436       | 0.6966       | 0.153*       |
| H13C | -0.0704     | 0.2944       | 0.5974       | 0.153*       |
| O1   | 1.0292 (2)  | 0.32081 (16) | 0.09966 (8)  | 0.0719 (4)   |
| O2   | 1.3810 (2)  | 0.30846 (17) | 0.17469 (9)  | 0.0749 (4)   |
| O3   | 0.6500 (2)  | 0.1095 (2)   | 0.56403 (9)  | 0.0889 (5)   |
| O4   | 0.3154 (2)  | 0.3006 (2)   | 0.52876 (8)  | 0.0778 (4)   |
| S1   | 1.20983 (6) | 0.22766 (5)  | 0.16117 (3)  | 0.05254 (19) |

Atomic displacement parameters ( $\text{\AA}^2$ )

|     | $U^{11}$    | $U^{22}$    | $U^{33}$    | $U^{12}$     | $U^{13}$    | $U^{23}$     |
|-----|-------------|-------------|-------------|--------------|-------------|--------------|
| C1  | 0.0426 (8)  | 0.0583 (10) | 0.0721 (10) | -0.0123 (7)  | 0.0000 (7)  | -0.0144 (8)  |
| C2  | 0.0466 (9)  | 0.0661 (12) | 0.0762 (11) | 0.0015 (8)   | 0.0056 (8)  | -0.0106 (9)  |
| C3  | 0.0706 (11) | 0.0474 (9)  | 0.0635 (9)  | -0.0046 (8)  | 0.0109 (8)  | -0.0111 (7)  |
| C4  | 0.0729 (12) | 0.0618 (12) | 0.0873 (12) | -0.0260 (10) | 0.0008 (10) | -0.0236 (10) |
| C5  | 0.0462 (8)  | 0.0578 (10) | 0.0782 (11) | -0.0109 (8)  | -0.0021 (8) | -0.0140 (8)  |
| C6  | 0.0428 (7)  | 0.0419 (8)  | 0.0454 (7)  | -0.0087 (6)  | 0.0009 (6)  | -0.0020 (6)  |
| C7  | 0.0576 (9)  | 0.0475 (9)  | 0.0494 (8)  | -0.0075 (7)  | -0.0029 (7) | -0.0041 (6)  |
| C8  | 0.0565 (9)  | 0.0496 (9)  | 0.0516 (8)  | -0.0064 (7)  | -0.0025 (7) | -0.0074 (7)  |
| C9  | 0.0590 (10) | 0.0562 (10) | 0.0548 (9)  | -0.0099 (8)  | -0.0007 (7) | -0.0047 (7)  |
| C10 | 0.0572 (10) | 0.0603 (11) | 0.0552 (9)  | -0.0060 (8)  | -0.0003 (7) | -0.0046 (8)  |
| C11 | 0.0595 (10) | 0.0586 (11) | 0.0615 (10) | -0.0110 (8)  | 0.0010 (8)  | -0.0047 (8)  |
| C12 | 0.0948 (15) | 0.0925 (16) | 0.0674 (11) | -0.0119 (13) | 0.0204 (11) | 0.0019 (11)  |
| C13 | 0.0908 (16) | 0.122 (2)   | 0.0886 (15) | -0.0233 (15) | 0.0273 (13) | -0.0240 (15) |
| O1  | 0.0726 (8)  | 0.0597 (8)  | 0.0600 (7)  | 0.0108 (6)   | -0.0020 (6) | 0.0051 (5)   |
| O2  | 0.0824 (9)  | 0.0565 (8)  | 0.0942 (9)  | -0.0331 (7)  | 0.0119 (7)  | -0.0141 (7)  |
| O3  | 0.0740 (9)  | 0.1009 (12) | 0.0691 (8)  | -0.0048 (8)  | -0.0020 (7) | 0.0225 (7)   |

|    |            |             |            |               |              |               |
|----|------------|-------------|------------|---------------|--------------|---------------|
| O4 | 0.0705 (8) | 0.0792 (10) | 0.0640 (7) | 0.0013 (7)    | 0.0155 (6)   | 0.0001 (7)    |
| S1 | 0.0557 (3) | 0.0399 (3)  | 0.0553 (3) | -0.00635 (19) | 0.00194 (18) | -0.00117 (17) |

*Geometric parameters (Å, °)*

|             |             |               |             |
|-------------|-------------|---------------|-------------|
| C1—C2       | 1.378 (2)   | C8—H8         | 0.9300      |
| C1—C6       | 1.378 (2)   | C9—C10        | 1.326 (2)   |
| C1—H1       | 0.9300      | C9—H9         | 0.9300      |
| C2—C3       | 1.359 (2)   | C10—C11       | 1.483 (2)   |
| C2—H2       | 0.9300      | C10—H10       | 0.9300      |
| C3—C4       | 1.363 (3)   | C11—O3        | 1.195 (2)   |
| C3—H3       | 0.9300      | C11—O4        | 1.319 (2)   |
| C4—C5       | 1.389 (3)   | C12—C13       | 1.451 (3)   |
| C4—H4       | 0.9300      | C12—O4        | 1.469 (2)   |
| C5—C6       | 1.383 (2)   | C12—H12A      | 0.9700      |
| C5—H5       | 0.9300      | C12—H12B      | 0.9700      |
| C6—S1       | 1.7595 (15) | C13—H13A      | 0.9600      |
| C7—C8       | 1.320 (2)   | C13—H13B      | 0.9600      |
| C7—S1       | 1.7434 (15) | C13—H13C      | 0.9600      |
| C7—H7       | 0.9300      | O1—S1         | 1.4294 (12) |
| C8—C9       | 1.452 (2)   | O2—S1         | 1.4328 (13) |
| C2—C1—C6    | 119.12 (15) | C8—C9—H9      | 117.4       |
| C2—C1—H1    | 120.4       | C9—C10—C11    | 119.36 (17) |
| C6—C1—H1    | 120.4       | C9—C10—H10    | 120.3       |
| C3—C2—C1    | 120.67 (15) | C11—C10—H10   | 120.3       |
| C3—C2—H2    | 119.7       | O3—C11—O4     | 123.62 (16) |
| C1—C2—H2    | 119.7       | O3—C11—C10    | 124.36 (17) |
| C2—C3—C4    | 120.76 (17) | O4—C11—C10    | 112.01 (15) |
| C2—C3—H3    | 119.6       | C13—C12—O4    | 108.24 (19) |
| C4—C3—H3    | 119.6       | C13—C12—H12A  | 110.1       |
| C3—C4—C5    | 119.78 (17) | O4—C12—H12A   | 110.1       |
| C3—C4—H4    | 120.1       | C13—C12—H12B  | 110.1       |
| C5—C4—H4    | 120.1       | O4—C12—H12B   | 110.1       |
| C6—C5—C4    | 119.32 (16) | H12A—C12—H12B | 108.4       |
| C6—C5—H5    | 120.3       | C12—C13—H13A  | 109.5       |
| C4—C5—H5    | 120.3       | C12—C13—H13B  | 109.5       |
| C1—C6—C5    | 120.34 (15) | H13A—C13—H13B | 109.5       |
| C1—C6—S1    | 119.90 (12) | C12—C13—H13C  | 109.5       |
| C5—C6—S1    | 119.71 (11) | H13A—C13—H13C | 109.5       |
| C8—C7—S1    | 123.19 (13) | H13B—C13—H13C | 109.5       |
| C8—C7—H7    | 118.4       | C11—O4—C12    | 115.52 (15) |
| S1—C7—H7    | 118.4       | O1—S1—O2      | 119.36 (8)  |
| C7—C8—C9    | 120.95 (15) | O1—S1—C7      | 108.53 (8)  |
| C7—C8—H8    | 119.5       | O2—S1—C7      | 107.09 (8)  |
| C9—C8—H8    | 119.5       | O1—S1—C6      | 109.53 (7)  |
| C10—C9—C8   | 125.22 (16) | O2—S1—C6      | 108.57 (7)  |
| C10—C9—H9   | 117.4       | C7—S1—C6      | 102.40 (7)  |
| C6—C1—C2—C3 | -0.3 (3)    | O3—C11—O4—C12 | 4.5 (3)     |

|               |              |                |              |
|---------------|--------------|----------------|--------------|
| C1—C2—C3—C4   | 0.9 (3)      | C10—C11—O4—C12 | -176.12 (16) |
| C2—C3—C4—C5   | -0.5 (3)     | C13—C12—O4—C11 | -171.7 (2)   |
| C3—C4—C5—C6   | -0.4 (3)     | C8—C7—S1—O1    | -4.46 (18)   |
| C2—C1—C6—C5   | -0.7 (2)     | C8—C7—S1—O2    | 125.64 (16)  |
| C2—C1—C6—S1   | 176.98 (12)  | C8—C7—S1—C6    | -120.23 (16) |
| C4—C5—C6—C1   | 1.0 (2)      | C1—C6—S1—O1    | 144.75 (14)  |
| C4—C5—C6—S1   | -176.61 (14) | C5—C6—S1—O1    | -37.59 (14)  |
| S1—C7—C8—C9   | -177.71 (12) | C1—C6—S1—O2    | 12.83 (15)   |
| C7—C8—C9—C10  | 176.02 (18)  | C5—C6—S1—O2    | -169.51 (12) |
| C8—C9—C10—C11 | -179.12 (15) | C1—C6—S1—C7    | -100.21 (14) |
| C9—C10—C11—O3 | -8.3 (3)     | C5—C6—S1—C7    | 77.46 (14)   |
| C9—C10—C11—O4 | 172.26 (17)  |                |              |

*Hydrogen-bond geometry (Å, °)*

| <i>D</i> —H $\cdots$ <i>A</i>  | <i>D</i> —H | H $\cdots$ <i>A</i> | <i>D</i> $\cdots$ <i>A</i> | <i>D</i> —H $\cdots$ <i>A</i> |
|--------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| C7—H7 $\cdots$ O3 <sup>i</sup> | 0.93        | 2.32                | 3.212 (2)                  | 161                           |

Symmetry code: (i)  $-x+2, -y, -z+1$ .